

## Piperazine-1,4-dium pyridine-2,3-dicarboxylate methanol monosolvate<sup>1</sup>

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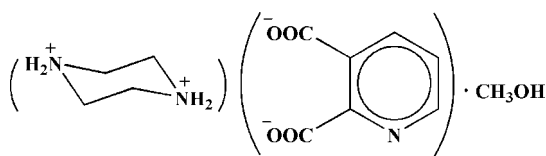
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Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.001$  Å;  $R$  factor = 0.034;  $wR$  factor = 0.085; data-to-parameter ratio = 19.7.

The title solvated molecular salt,  $\text{C}_4\text{H}_{12}\text{N}_2^{2+} \cdot \text{C}_7\text{H}_3\text{NO}_4^{2-} \cdot \text{CH}_3\text{OH}$  or  $(\text{pipzH}_2)(\text{py-2,3-dc}) \cdot \text{MeOH}$ , was prepared by the reaction of pyridine-2,3-dicarboxylic acid ( $\text{py-2,3-dcH}_2$ ) and piperazine ( $\text{pipz}$ ) in methanol ( $\text{MeOH}$ ) as solvent. One of the two carboxylate groups of the acid fragment is nearly perpendicular to the pyridine ring and the other is almost in its plane [ $\text{C}-\text{C}-\text{O}$  torsion angles =  $-85.50$  (11) and  $88.07$  (11)° and  $\text{N}-\text{C}-\text{C}-\text{O}$  torsion angles =  $-176.31$  (8) and  $5.41$  (13)°]. In the crystal, the components are linked by  $\text{O}-\text{H} \cdots \text{O}$ ,  $\text{N}-\text{H} \cdots \text{O}$  and  $\text{C}-\text{H} \cdots \text{O}$  hydrogen bonds, generating a three-dimensional network.

### Related literature

For similar ion pairs, see: Aghabozorg, Manteghi & Ghadermazi (2008); Aghabozorg, Manteghi & Sheshmani (2008). For related metal complexes, see: Barszcz *et al.* (2010); Li & Li (2004).



### Experimental

#### Crystal data

$\text{C}_4\text{H}_{12}\text{N}_2^{2+} \cdot \text{C}_7\text{H}_3\text{NO}_4^{2-} \cdot \text{CH}_3\text{O}$

$M_r = 285.30$

Monoclinic,  $P2_1/n$

$a = 8.2541$  (6) Å

$b = 11.8988$  (8) Å

$c = 13.8197$  (9) Å

$\beta = 90.288$  (2)°  
 $V = 1357.27$  (16) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation

$\mu = 0.11$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.25 \times 0.20 \times 0.10$  mm

#### Data collection

Bruker SMART APEXII diffractometer  
 16044 measured reflections

3579 independent reflections  
 3189 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.023$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$   
 $wR(F^2) = 0.085$   
 $S = 1.03$   
 3579 reflections

182 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.42$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N2}-\text{H2A} \cdots \text{O4}^{\text{I}}$	0.90	1.74	2.6257 (11)	168
$\text{N2}-\text{H2B} \cdots \text{O2}$	0.90	1.89	2.7274 (11)	155
$\text{N3}-\text{H3A} \cdots \text{O1}^{\text{II}}$	0.90	1.85	2.7379 (11)	169
$\text{N3}-\text{H3B} \cdots \text{O3}^{\text{III}}$	0.90	1.86	2.7393 (11)	166
$\text{O5}-\text{H5A} \cdots \text{O1}$	0.85	1.84	2.6867 (10)	171
$\text{C3}-\text{H3} \cdots \text{O5}^{\text{IV}}$	0.95	2.41	3.3163 (13)	159

Symmetry codes: (i)  $-x, -y + 2, -z$ ; (ii)  $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (iii)  $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (iv)  $x - 1, y, z$ .

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINTE* (Bruker, 2005); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009) and *Mercury* (Macrae *et al.*, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: OM2416).

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<sup>1</sup> In memory of our great professor, Dr Hossein Aghabozorg, who passed away recently.